PHY 555: Solid-state Physics I

Homework #6-7 Due: 12/09/2024

Conceptual

- **1.** *5 points* Explain what the one-electron approximation is and how Hartree-Fock and density-functional theory go beyond it.
- **2.** *5 points* Compare and contrast Hartree-Fock and density-functional theory in terms of the following aspects:
 - (a) General philosophy for addressing the many-electron problem.
 - (**b**) The resulting single-electron problem to solve.
 - (c) The physical interpretation of the auxiliary single-particle orbitals and eigenvalues.
- 3. Explain the adiabatic Born-Oppenheimer approximation used in calculating lattice dynamics.

Analytical

- **4.** 20 *points* Pretend that electrons are distinguishable particles, so that an appropriate many-electron wavefunction is the product of spin orbitals: $\Psi(\mathbf{r}_1\sigma_1, ..., \mathbf{r}_N\sigma_N) = \prod_i^N \psi_i(\mathbf{r}_i\sigma_i)$. For this many-body wavefunction ansatz, derive the single particle equations for ψ that minimize the many-body energy (like we did in the Hartree-Fock approach).
- **5.** *20 points* Previously, we have discussed the free electron gas via the Sommerfeld model, which assumed that the electrons were not interacting. We will now treat the same problem with Hartree-Fock theory.
 - (a) In order for the system to be charge neutral, we include a uniform neutralizing positive background charge with the same density as the electrons (referred to as the "Jellium model"). Show that in this model the direct Coulomb interaction and external potential (i.e., the interaction fo the electrons with the positive background, neglect the interaction with the positive background and itself) cancel so that the Fock operator only includes the kinetic energy and exchange.
 - (b) Show that plane waves are eigenstates of the Fock operator. To simplify the eigenvectors, use the fact that the Fourier transform of $e^2/|\mathbf{r} \mathbf{r}'|$
 - (c) What is the ground-state energy of the free electron gas in Hartree-Fock theory as a function of r_s ?

6. 20 points We have discussed plane waves and atomic orbitals for performing calculations of the electronic structure of solids and materials. In addition, *gaussians* are another common basis set, and useful for other calculations in solids. The benefit of gaussians as a basis set is that they are localized functions, so do not require pseudopotentials like plane waves, and integrals are easier to calculate than atomic orbitals. For example, one of the useful property about gaussians is the *gaussian product theorem* (GPT) that states that the product of two gaussians is also a gaussian, centered at the "center of gravity" of the two original gaussians. I.e., for $\chi_a(\mathbf{r}) = e^{-\alpha(\mathbf{r}-\mathbf{A})^2}$ and $\chi_b(\mathbf{r}) = e^{-\beta(\mathbf{r}-\mathbf{B})^2}$,

$$\chi_a(\mathbf{r})\chi_b(\mathbf{r}) = e^{-\frac{\alpha\beta}{\alpha+\beta}(\mathbf{A}-\mathbf{B})^2} e^{-(\alpha+\beta)(\mathbf{r}-\mathbf{P})^2}$$
(1)

where $\mathbf{P} = (\alpha \mathbf{A} + \beta \mathbf{B}) / (\alpha + \beta)$. We will use this and other properties of Gaussians in the next several problems.

- **7.** 20 *points* Consider a basis set of gaussians of the form $\phi_i(\mathbf{r}) = A_i e^{\alpha_i |\mathbf{r} \mathbf{R}_i|^2}$.
 - (a) Using the GPT [Eq. (1)], show that we can calculate the overlap between gaussians, i.e., $S_{ij} = \int \phi_i(\mathbf{r})\phi_j(\mathbf{r})d\mathbf{r}$ with the simple relation

$$S_{ij} = A_i A_j e^{-\frac{\alpha_i \alpha_j}{\alpha_i + \alpha_j} (\mathbf{R}_i - \mathbf{R}_j)^2} \left(\frac{\pi}{\alpha_i + \alpha_j}\right)^{3/2}$$
(2)

Note that we neglect the complex conjugation since all gaussians considered here will be real.

(b) In general Coulomb matrix elements, for V_{ext} and the two electron integrals, are tricky to calculate. This is because we need to be careful integrating over divergences of the form $1/|\mathbf{r} - \mathbf{r}'|$ when $\mathbf{r} = \mathbf{r}'$. With gaussians, there is an elegant way to evaluate these matrix elements, utilizing the integral transformation

$$\frac{1}{|\mathbf{r} - \mathbf{r}_C|} = \int_{-\infty}^{\infty} e^{-|\mathbf{r} - \mathbf{r}_C|^2 t^2} dt$$
(3)

Consider a gaussian charge density $\rho_i(\mathbf{r}) = (\alpha_i / \pi)^{3/2} \exp(-\alpha_i |\mathbf{r} - \mathbf{R}_i|)$, which could correspond to a basis function, or product of basis functions (both are gaussians!). Show using the result of (a) that $\rho_i(\mathbf{r})$ is normalized to unity. Then, using Eq. (3) and Eq. (1), show that the electrostatic potential at point \mathbf{R}_C , i.e.,

$$V_i(\mathbf{R}_C) = \int \frac{\rho_i(\mathbf{r})}{|\mathbf{r} - \mathbf{R}_C|} d\mathbf{r}$$
(4)

can be written as

$$V_i(\mathbf{R}_C) = \frac{1}{|\mathbf{R}_i - \mathbf{R}_C|} \operatorname{erf}\left(\sqrt{\alpha_i |\mathbf{R}_i - \mathbf{R}_C|^2}\right),\tag{5}$$

where erf is the error function and we use atomic units throughout so $m_e = e = \hbar = 1$.

(c) Using the result of (b) and Eq. (1), show that an arbitrary Coulomb matrix element between gaussians

$$U_{ijkl} = \int \int \frac{\phi_i(\mathbf{r}_1)\phi_j(\mathbf{r}_1)\phi_k(\mathbf{r}_2)\phi_l(\mathbf{r}_2)}{|\mathbf{r}_1 - \mathbf{r}_2|} d\mathbf{r}_1 d\mathbf{r}_2$$
(6)

can be written as

$$U_{ijkl} = \frac{S_{ij}S_{kl}}{|\mathbf{R}_i - \mathbf{R}_C|} \operatorname{erf}\left(\sqrt{\alpha_i|\mathbf{R}_i - \mathbf{R}_C|^2}\right)$$
(7)

8. 20 *points* An important contribution to the energy of ionic materials is the sum of electrostatic interactions between oppositely charged ions

Computational

9. In this problem, we consider a simple diatomic molecule made up of a hydrogen atom and a helium atom (H-He) with two electrons. We set the location of H at $\mathbf{R}_{\rm H} = (0,0,0)$ and He is $\mathbf{R}_{\rm He} = (1.5117,0,0)$ Ha. We use a basis set made up of two Gaussians:

$$\phi_1(\mathbf{r}) = 0.3696e^{-0.4166|\mathbf{r} - \mathbf{R}_{\rm H}|^2} \tag{8}$$

$$\phi_2(\mathbf{r}) = 0.5881 e^{-0.7739|\mathbf{r} - \mathbf{R}_{\text{He}}|^2} \tag{9}$$

We will neglect spin for now, and will use atomic units throughout so $m_e = e = \hbar = 1$.

(a) Calculate the kinetic energy matrix elements with the basis functions:

$$T_{ij} = \int \phi_i(\mathbf{r}) \left(-\frac{1}{2}\nabla^2\right) \phi_j(\mathbf{r}) d\mathbf{r}.$$
 (10)

(b) Use your results from problem 6 to calculate the matrix elements of the external potential

$$V_{ij} = -\int \phi_i(\mathbf{r}) \frac{Z_{\rm H}}{|\mathbf{r} - \mathbf{R}_{\rm H}|} \phi_j(\mathbf{r}) d\mathbf{r} - \int \phi_i(\mathbf{r}) \frac{Z_{\rm He}}{|\mathbf{r} - \mathbf{R}_{\rm He}|} \phi_j(\mathbf{r}) d\mathbf{r}.$$
 (11)

(c) Use your results from problem 6 to calculate the Coulomb and exchange matrix elements

$$U_{ijkl} = \int \int \frac{\phi_i(\mathbf{r}_1)\phi_j(\mathbf{r}_1)\phi_k(\mathbf{r}_2)\phi_l(\mathbf{r}_2)}{|\mathbf{r}_1 - \mathbf{r}_2|} d\mathbf{r}_1 d\mathbf{r}_2.$$
(12)

- (d) Calculate the nuclear-nuclear repulsion energy:
- **10.** Now we have all of the information to solve for the energy of the molecule using Hartree-Fock (HF). directly solving the HF equation is challenging, especially because of the exchange term, which makes it an integro-differential equation. For "closed shell" systems where we can neglect the spin part (as we do here), we can actually express the equation in a form much easier to solve:

$$FC = SC\epsilon.$$
(13)

In this equation, **S** is the overlap matrix between basis functions discussed in problem **6**(**a**); **C** is the matrix of expansion coefficients for the basis elements, i.e., the wavefunctions solving the HF equations will be given by is given by $\psi_i = \sum_i C_{ij}\phi_i$; and **F** is the Fock matrix with elements

$$F_{ij} = T_{ij} + V_{ij} + \sum_{kl} P_{kl} (U_{ijkl} - \frac{1}{2} U_{ilkj}),$$
(14)

where P_{kl} are the elements of the "charge density matrix." They are related to the expansion coefficient via $P_{kl} = 2\sum_{i}^{N/2} C_{ki}C_{li}^*$ where *N* is the number of electrons in the system (in our case there will be two). It is called the charge-density matrix because the total charge density of the system can be written as $\rho(\mathbf{r}) = \sum_{ij} P_{ij}\phi_i(\mathbf{r})\phi_i^*(\mathbf{r})$. Perform the following steps to solve for the energy of the H-He molecule using Hartree-Fock: